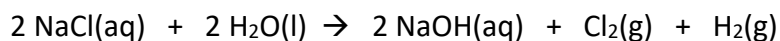


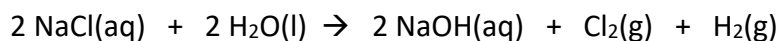
Warm-Up #34 Name: _____ Date: _____



1) How many moles of hydrogen gas will be produced when 5.55 moles of sodium chloride react?

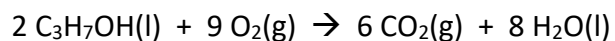
2) If 23.0 grams of chlorine gas were produced, how many grams of sodium chloride reacted with water?

Warm-Up #34 Name: _____ Date: _____

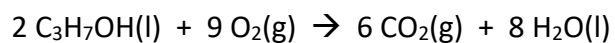


1) How many moles of hydrogen gas will be produced when 5.55 moles of sodium chloride react?

2) If 23.0 grams of chlorine gas were produced, how many grams of sodium chloride reacted with water?



- 3) How many liters of $\text{CO}_2(\text{g})$ will be produced by the complete reaction of 3.7×10^{24} molecules of $\text{C}_3\text{H}_7\text{OH}(\text{l})$?
- 4) How many molecules of oxygen gas are required to produce 44 grams of water?



- 3) How many liters of $\text{CO}_2(\text{g})$ will be produced by the complete reaction of 3.7×10^{24} molecules of $\text{C}_3\text{H}_7\text{OH}(\text{l})$?
- 4) How many molecules of oxygen gas are required to produce 44 grams of water?