

M = molecular = use prefixes

I = ionic = no prefixes

Warm-Up #25

Name: \_\_\_\_\_

Date: \_\_\_\_\_

	Name	I/M	Formula
1.	diphosphorus heptoxide	M	$P_2O_7$
2.	carbon monoxide	M	CO
3.	barium nitride	I	$Ba_3N_2$
4.	Potassium carbonate	I	$K^{1+} (CO_3)^{2-}$ $K_2CO_3$
5.	Sulfur <u>hexa</u> fluoride	M	$SF_6$
6.	Ammonium phosphate	I	$(NH_4)^{1+} (PO_4)^{3-}$

