

	Name	I/M	Formula
1.	Trisulfur octabromide NON NON	M	S_3Br_8
2.	Nickel (III) chloride ³⁺ Metal ¹⁻ Non	I	$NiCl_3$
3.	Cadmium hydroxide ²⁺ Metal ^{OH¹⁻} poly ion	I	$Cd(OH)_2$
4.	lead(II) carbonate ²⁺ ²⁻ poly ion	I	$PbCO_3$
5.	Silicon tetrafluoride M	M	SiF_4 NON NON
6.	Phosphorus pentahydride M	M	PH_5 NON NON