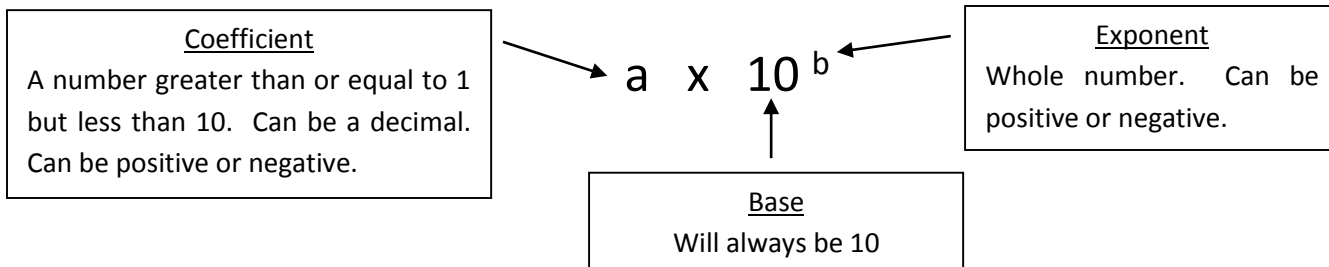


Scientific Notation

When numbers are extremely small or large it is easier to rewrite them in scientific notation.

Numbers are written in scientific notation using the following format:



Steps to convert into scientific notation:

1. Move the decimal point so that the coefficient is a number greater than or equal to 1 but less than 10.
2. Write the coefficient $\times 10$
3. The exponent is the number of places the decimal point was moved.
 - a. If the decimal point was moved to the right, then the exponent is negative.
 - b. If the decimal point was moved to the left, then the exponent is positive.

Examples:

If the number is **less than one**, the decimal point is moved to the **right** and the **exponent will be negative**.

$$0.0043 \longrightarrow 0.0043 \longrightarrow 4.3 \times 10^{-3}$$

3 places

If the number is **greater than one**, the decimal point is moved to the **left** and the **exponent will be positive**.

$$15\,000 \longrightarrow 15\,000 \longrightarrow 1.5 \times 10^4$$

4 places

Practice Part A: Rewrite the following values in scientific notation.

- 1) 45 148 287 cm
- 2) 1 289 m
- 3) 0.000 004 293 nm
- 4) 602 000 000 000 000 000 000 000 atoms
- 5) 0.0075 L
- 6) 0.000 619 s

Practice Part B: Rewrite the following values in standard form.

- 7) 3.765×10^7 m
- 8) 4.0×10^{-8} g
- 9) 7.26×10^{-5} L
- 10) $2.889\,241 \times 10^4$ kg