

silver chloride



zinc oxide



calcium bromide



strontium fluoride

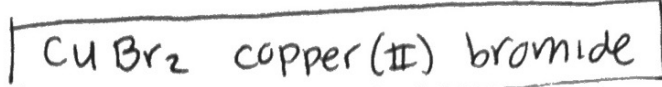


barium oxide

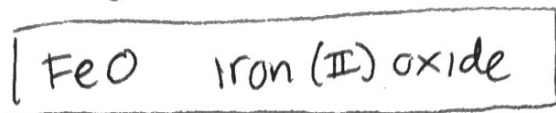


calcium chloride

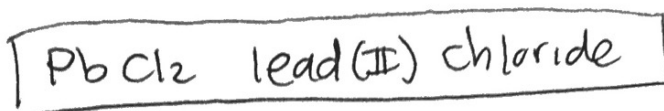
#1. a.  $\text{Cu}^{2+} \text{Br}^{-} \neq 0$   $\therefore$  crisscross & reduce



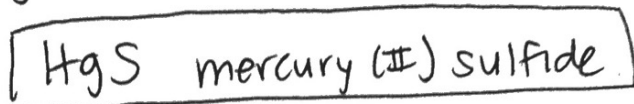
b.  $\text{Fe}^{2+} \text{O}^{2-} = 0$



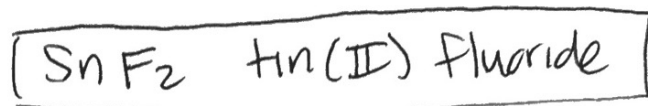
c.  $\text{Pb}^{2+} \text{Cl}^{-} \neq 0$   $\therefore$  crisscross & reduce



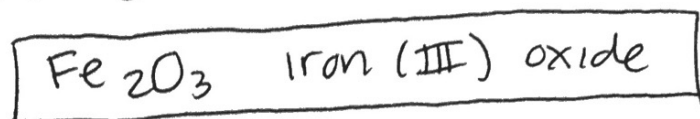
d.  $\text{Hg}^{2+} \text{S}^{2-} = 0$



e.  $\text{Sn}^{2+} \text{F}^{-} \neq 0$   $\therefore$  crisscross & reduce



f.  $\text{Fe}^{3+} \text{O}^{2-} \neq 0$   $\therefore$  crisscross & reduce

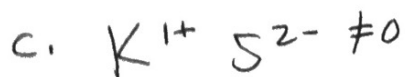
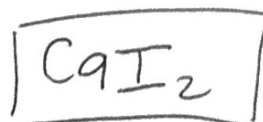
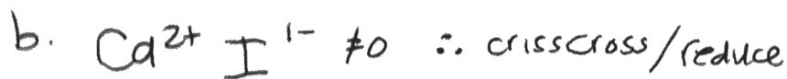
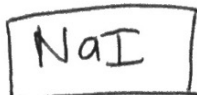
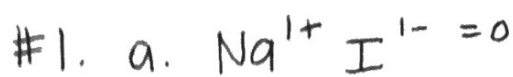


#2 a.  $\text{Cu}^{2+} \text{O}^{2-}$  copper (II) oxide

b.  $\text{Co}^{3+} \text{F}^{-}$  cobalt (III) fluoride

c.  $\text{Sn}^{4+} \text{I}^{-}$  tin (IV) iodide

d.  $\text{Fe}^{2+} \text{S}^{2-}$  iron (II) sulfide



$\therefore$  crisscross/reduce

