

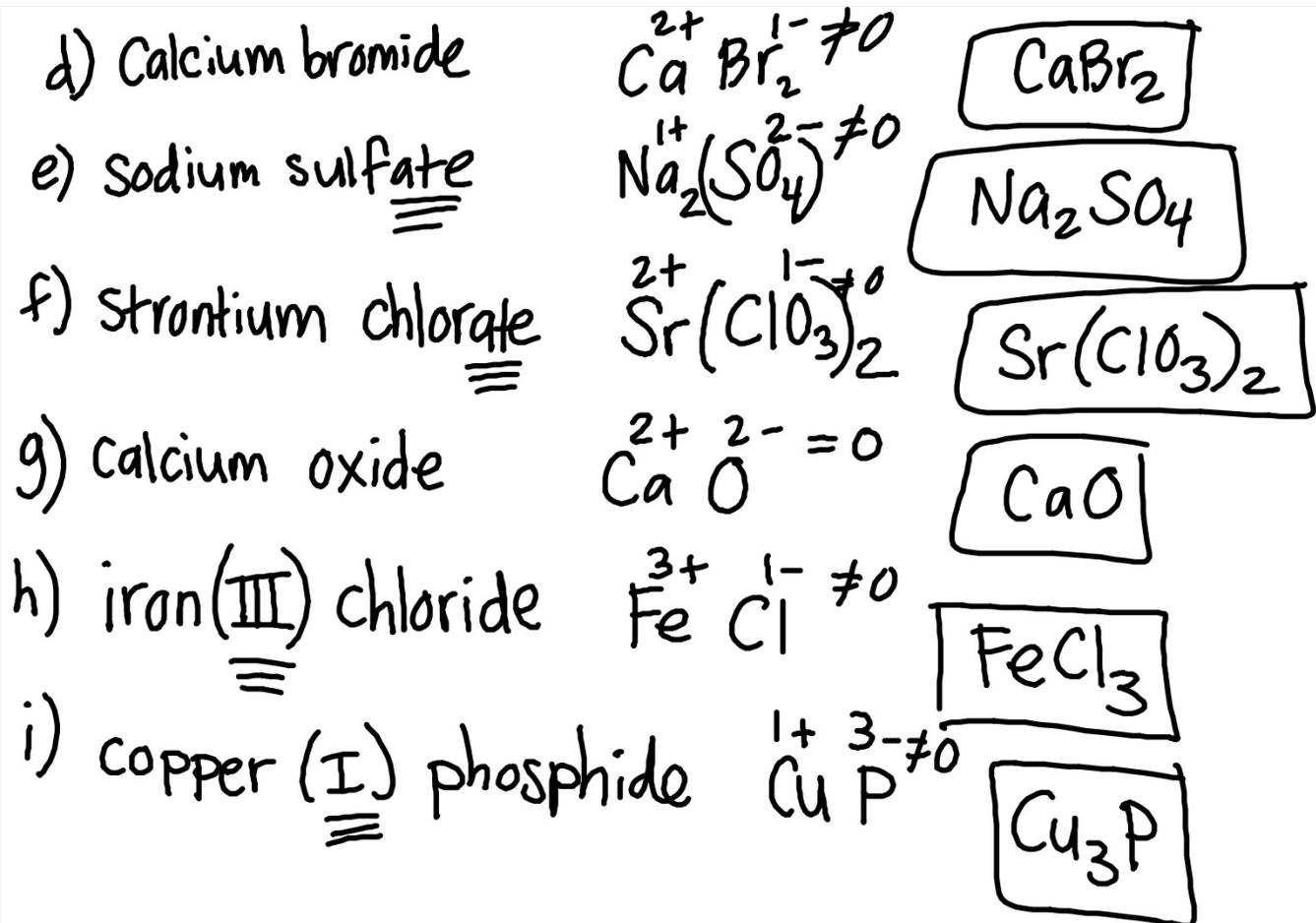
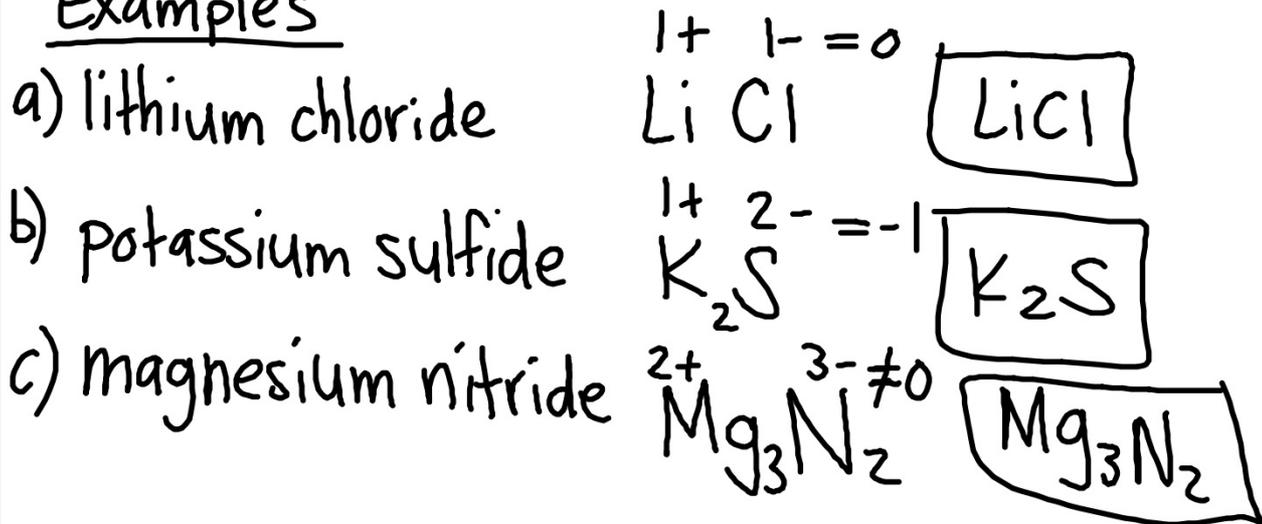
# Writing Ionic Formulas Notes

Ionic compound = metal + nonmetal

\* Ionic formulas must be neutral (no charge)

Sum of  $\oplus$  ions and  $\ominus$  ions must equal  $\emptyset$

## Examples





cobalt(II) bromide



chromium(III) nitride



silver nitrate



lead(IV) oxide