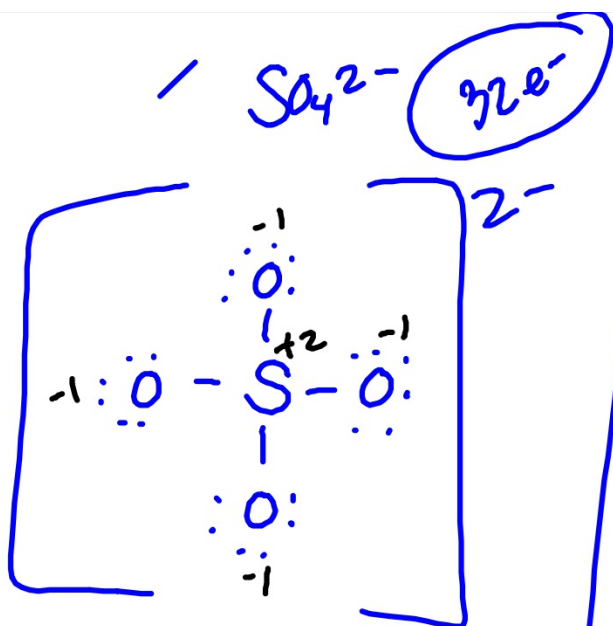
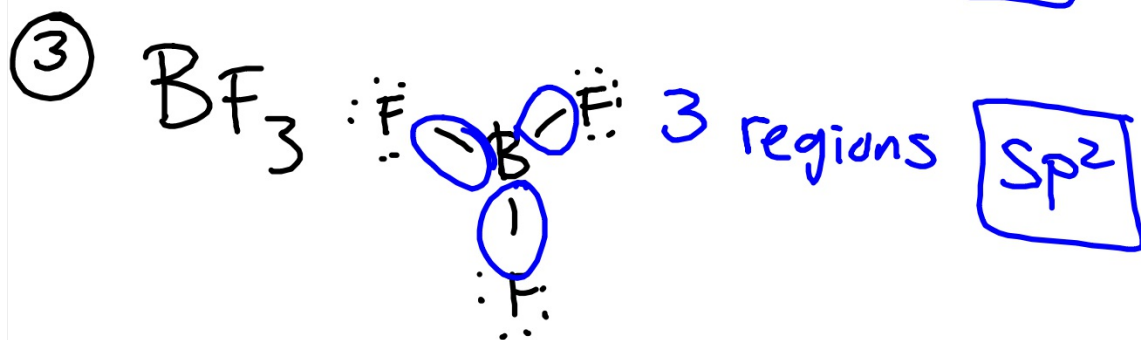
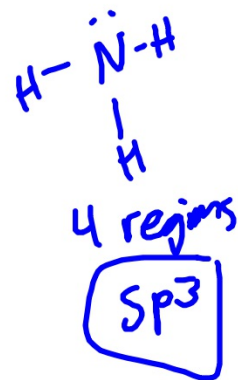
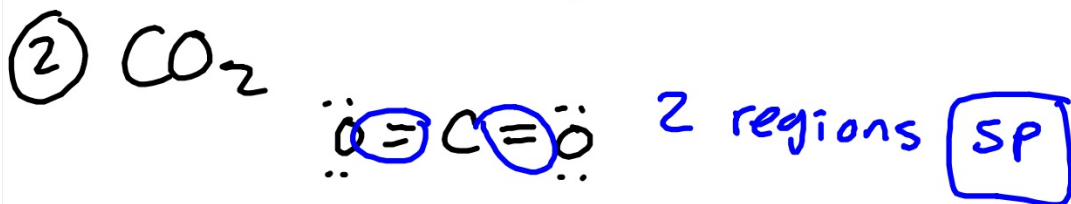
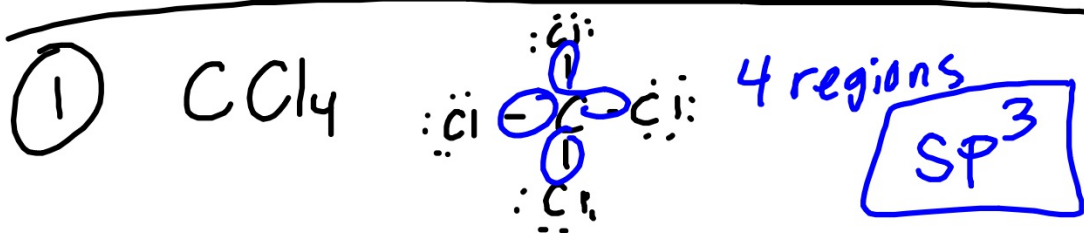
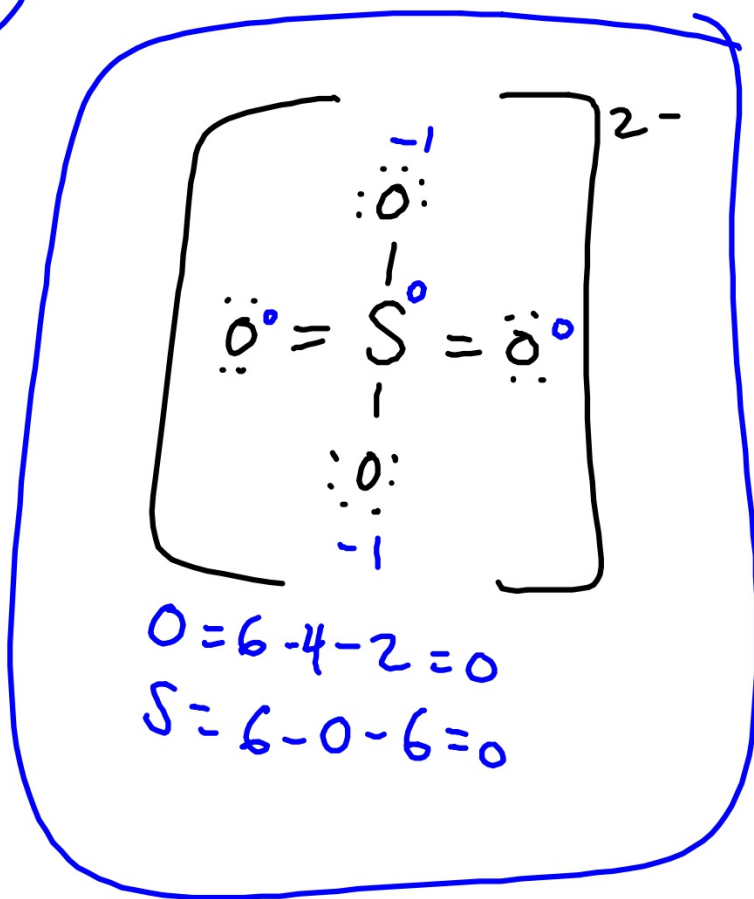


Determine the hybridization of the center atom.



$$O = 6 - 6 - 1 = -1$$

$$S = 6 - 0 - 4 = +2$$



$$O = 6 - 4 - 2 = 0$$

$$S = 6 - 0 - 6 = 0$$

$$\left[\begin{array}{c} \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \end{array} = S^{-2} = \begin{array}{c} \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \\ \ddot{0} \end{array} \right]^{2-}$$

$$S = 6 - 0 - 8 = -2$$