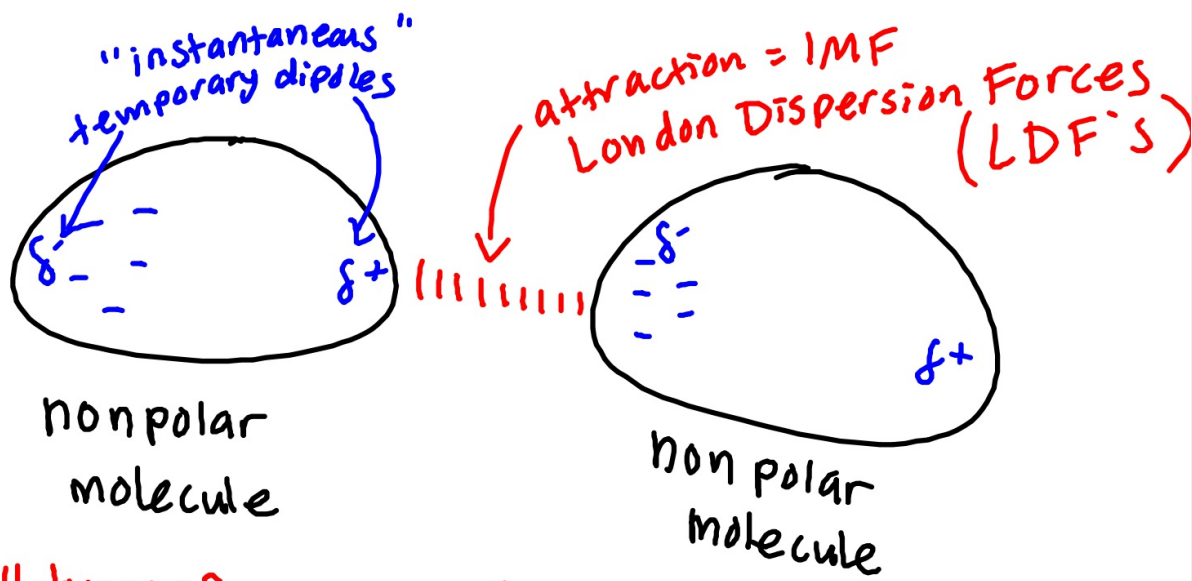


★ H-bonding requirement
must have H covalently bonded to
N, O, or F in same molecule



- ★ All types of molecules (polar + nonpolar) have LDF's.
 - ★ more e^- = easier to form temporary (instantaneous) dipoles = more polarizable = stronger LDF's
- think higher molar mass